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Page 2 of 8.  $\text{H}_3\text{O}^+(\text{aq}) + \text{NH}_3(\text{aq}) \rightarrow \text{H}_2\text{O}(\text{l}) + \text{NH}_4^+(\text{aq})$ . This involves transfer of a proton from the hydronium ion to the ammonia molecule to produce water and the ammonium ion. This makes  $\text{H}_3\text{O}^+$  an acid and  $\text{NH}_3$  a

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compare to the answer you obtained in Key Question. Based on the data in Model 1 and the table in Question 3, describe the relationship between: a.

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